

# INFLUENCE OF PORTLAND POZZOLANA CEMENT ON THE SERVICE LIFE OF REINFORCED CONCRETE UNDER CHLORIDE ATTACK

Bhaskar Sangoju<sup>1</sup>, Radhakrishna G. Pillai<sup>2</sup>, Ravindra Gettu<sup>2</sup>,  
B.H. Bharatkumar<sup>1</sup>, and Nagesh R. Iyer<sup>1</sup>

<sup>1</sup> *CSIR-Structural Engineering Research Centre (SERC), Council of Scientific and Industrial Research (CSIR) Campus, Chennai, INDIA*

<sup>2</sup> *Indian Institute of Technology Madras (IITM), Chennai, INDIA*

## ABSTRACT

This paper discusses the influence of blended cement on the service life of reinforced concrete (RC) structural components subjected to chloride-rich environments. The service life is assumed as the sum of the corrosion initiation and propagation periods. A comprehensive experimental programme was performed to obtain the chloride diffusion coefficient and corrosion current density that are used in the estimation of the corrosion initiation and propagation periods. The estimated service lives of ordinary portland cement (OPC) and portland pozzolana cement (PPC) concretes having thermo-mechanically treated steel reinforcement, when exposed to chloride environments, are presented. The results suggest that, under certain circumstances, the service life of an RC structure can double when PPC is used instead of OPC.

**Keywords:** Concrete, chloride induced corrosion, diffusion, service life.

## INTRODUCTION

The highly alkaline environment within the fresh and uncontaminated concrete protects the embedded reinforcing bars (rebars) from corrosion. Such protection is obtained due to the presence of a dense ferrous ( $\text{Fe}^{2+}$ ) or ferric ( $\text{Fe}^{3+}$ ) oxide layer (known as passive film) that prevents further oxidization of the rebar (Broomfield, 2006). In chloride rich environments, the chlorides from outside environment can diffuse, through the cover concrete, towards the steel reinforcement. The corrosion of the rebar initiates when the chloride concentration at or near the rebar surface reaches a minimum concentration, known as the critical chloride threshold concentration ( $C_{tc}$ ). The value of  $C_{tc}$  is an important parameter necessary for estimating the service life. This parameter exhibits significant variation and can depend on various factors (Richardson, 2002). The chloride-induced corrosion is catalytic in nature and is very difficult to stop, once initiated. The volume of the corrosion products or rust produced can be approximately 6 to 8 times more than the volume of the original steel

corroded. This volumetric expansion can create significant tensile stresses in the cover concrete, resulting in cracking and spalling of the cover concrete. Typically, a particular amount (say, 10%) of cracking/spalling or structural capacity loss will define the end of service life, unless adequate repair work is performed.

The prediction of service life is necessary to ensure long term safety and sustainability of RC structures subjected to chloride-induced corrosion. As shown in Figure 1, the overall service life of an RC structure exposed to chloride environments can be taken as the sum of the durations of corrosion initiation, corrosion propagation, and repair phases. It should be noted that the duration of the initiation phase is typically much longer than that of the propagation or repair phases.

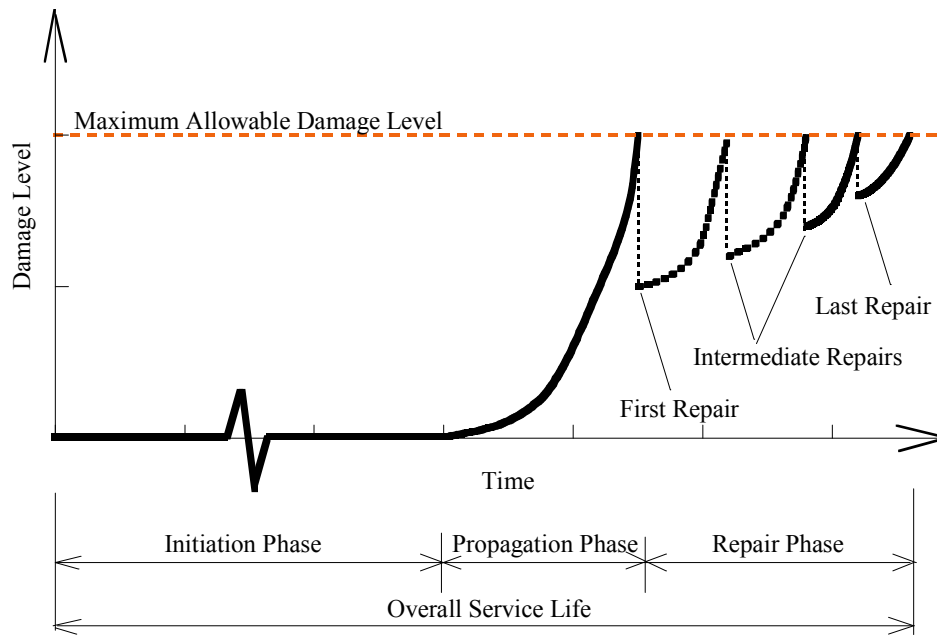


Figure 1. Different stages of deterioration in an RC structure (Pillai, 2003)

Many investigators have considered corrosion deterioration of RC in two stages: (i) initiation stage and (ii) propagation (Tuutti, 1982; Amey et al., 1998). In the initiation stage, the main process is the ingress of chloride ions towards the rebar. In this stage, the corrosion rate of the rebar is negligible, as long as the chloride content is below  $C_{ic}$  and the passivity of the reinforcement is maintained. The duration of the initiation stage could vary widely, depending on the aggressiveness of the temperature, humidity and chloride conditions of the environment, the quality of the cover concrete and the  $C_{ic}$  of the embedded steel. The propagation stage is defined as the time between the initiation of corrosion of the rebar and the cracking and spalling of the cover concrete. It has been reported that the rate of corrosion is uniform or constant (i.e., stable) at the earlier stages of the propagation stage, and can increase and vary (or become unstable) at later stages. This increase and variation in later stages is because the rate of ingress of moisture, chlorides and oxygen and/or the rate of removal of the existing dense and thick layer of corrosion products (which provides some protection against further corrosion) can increase when the cover concrete starts cracking. The corrosion process/progress at this later stage could differ from those occurring before the

cover concrete is cracked. Consequently, the deterioration stages of RC structures due to rebar corrosion can be classified into three, namely (Austroads, 2000): corrosion initiation, stable corrosion propagation and unstable/unpredictable corrosion propagation.

Many researchers have proposed different methodologies for predicting the service life when exposed to chloride ions (Ahmad et al., 1997; Anoop et al., 2002; ACI 365.1R, 2010; Life-365, 2010). However, in most literature, the service life is approximated to be only the length of the corrosion initiation phase (Tuutti, 1982; Bentur et al., 1998; Life-365, 2010). Such an approach may be too conservative in most cases because of (1) the excess reinforcement provided due to safety considerations and (2) the insignificant damage or cross-sectional loss of the reinforcement at the end of the corrosion initiation phase (Austroads, 2000). In this paper, the service life has been estimated as a sum of the corrosion initiation time ( $t_i$ ) and stable corrosion propagation time ( $t_{sp}$ ), as defined in Austroads (2000).

## EXPERIMENTAL PROGRAMME

Commercially available ordinary portland cement (OPC) and fly ash based portland pozzolana cement (PPC), and potable water were used in the mixes, along with river sand, and graded crushed granite (with maximum grain sizes of 20 mm and 12 mm in the ratio of 1.5:1). The ACI 211 (2010) guidelines were followed in the concrete mix design and a water to cement ratio (w/c) equal to 0.57 was used. The material quantities per cubic meter of the cement, fine and coarse aggregates were 300, 870 and 1056 kg, respectively. Specimens such as cubes of 150×150×150 mm for compressive strength test, cylinders of 100 mm diameter and 200 mm height for rapid chloride penetration test (RCPT) and chloride diffusion coefficient test were cast. The 28-day average compressive strength and charge passed in the RCPT (as per ASTM C 1202) for OPC and PPC concretes were 33.2 MPa and 2600 Coulombs; and 30.7 MPa and 861 Coulombs, respectively. A specially designed U-shaped RC specimen (Figure 2) with a TMT rebar of 12 mm diameter placed with 20 mm clear cover at bottom were subjected to accelerated chloride attack for studying the corrosion of the rebar in OPC and PPC concretes (see Bhaskar et al., 2011, for test details).

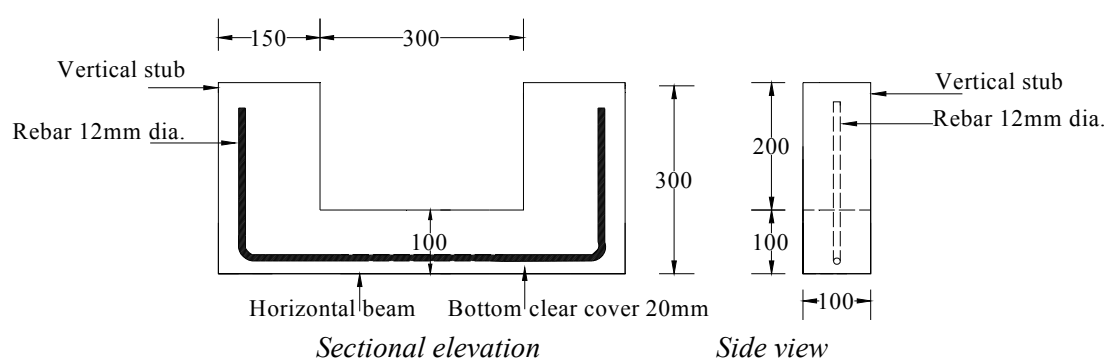


Figure 2. Schematic of the U-shaped corrosion test specimen

## Evaluation of chloride diffusion coefficient ( $D_c$ )

The chloride diffusivity is a parameter that reflects the resistance of the concrete to chloride penetration. In order to evaluate the apparent diffusion coefficient in the present study, tests have been carried out as per the Nordtest Method (NT Build 355, 1997). In this method, two

reservoirs separated by the concrete cylinder specimen are filled with 0.25 N sodium chloride (NaCl) solution and 0.25 N sodium hydroxide (NaOH) solution, and a constant voltage of 12 V DC is applied across the specimen. The chloride ions migrate from the NaCl solution reservoir, through the concrete specimen, to the NaOH solution reservoir. The chloride flux through the specimen is determined by using the increase in chloride concentration with time in the NaOH solution after steady state is achieved. For this, 50 ml of NaOH solution is taken from the reservoir at different intervals and tested for  $\text{Cl}^-$  concentration using standard volumetric titration method. A curve is drawn between chloride concentration,  $c$  (in  $\text{mmol/cm}^3$ ) and time,  $t$  (in days). For a constant voltage, the chloride concentration in the NaOH solution increases gradually and linearly with time. The slope,  $dc/dt$ , over the linear portion is determined and  $D_c$  value has been calculated as per the methodology proposed by Gjrv (1994), and Zhang and Gjrv (1994). The test is stopped when constant chloride concentration is observed over a period of time, i.e.,  $dc/dt = 0$  (Prabakar et al., 2010). The chloride diffusion coefficient,  $D_c$ , for OPC and PPC concretes have been found to be  $7.67 \times 10^{-13}$  and  $3.92 \times 10^{-13} \text{m}^2/\text{sec}$  at 6 months, respectively. The lower value of  $D_c$  for PPC concrete can be attributed to the pozzolanic reaction between fly ash and  $\text{Ca}(\text{OH})_2$  to form secondary calcium hydrates, causing pore size refinement (Saraswathy and Song, 2007).

## SERVICE LIFE OF STRUCTURES

The successful estimation of the initiation, stable propagation and unstable propagation periods is the basis for service life prediction. If all these periods are known within reasonable doubt, then the service life could be estimated. It is seen that different researchers have different opinions of what should be defined as the service life (Austroads, 2000; ACI 365.1R, 2010; Life-365, 2010). In the present study, it is assumed that the service life ( $t_s$ ) is the sum of the corrosion initiation period ( $t_i$ ) and stable propagation period ( $t_{sp}$ ) and expressed as follows (Austroads, 2000):

$$t_s = t_i + t_{sp} \quad (1)$$

### Initiation period ( $t_i$ )

Based on Fick's 2<sup>nd</sup> law of diffusion, many researchers have used the following closed-form solution for predicting  $t_i$  when the concrete is exposed to chloride environments (ACI 365.1R, 2010; Life-365, 2010):

$$C(x,t) = C_0 \left[ 1 - \text{erf} \left( \frac{x}{2\sqrt{(D_c t_i)}} \right) \right] \quad (2)$$

where  $C(x,t)$  is chloride concentration at depth  $x$  (or  $C_{ic}$  in this case) after time  $t_i$ ;  $C_0$  is the chloride concentration at the concrete surface;  $D_c$  is the apparent chloride diffusion coefficient;  $\text{erf}$  is the Gaussian error function, and  $x$  is the clear depth of rebar from concrete surface or clear cover thickness.

Once the values of  $C_{ic}$ ,  $C_0$  and  $D_c$  are available, the time for  $C(x,t)$  at the depth of the rebar to reach  $C_{ic}$  gives  $t_i$ . In reality, the values  $C_{ic}$ ,  $C_0$  and  $D_c$  are random variables. The value of  $C_{ic}$  depends upon various factors that include the microstructure and metallurgical parameters of

the rebar, the complex microstructure of the surrounding concrete, the pH of the concrete pore solution, binding capacity of cement, the local environmental characteristics, and the test procedures used to evaluate this parameter (Alonso et al., 2000; Pillai and Trejo, 2005; Angst et al., 2009). A higher threshold would lead to a longer predicted life and vice-versa, provided the other conditions are the same. Therefore, the threshold value should be carefully selected. In the present study, the  $C_{tc}$  value is assumed as 0.4%, by weight of cement, which is the most commonly used value for the estimation of  $t_i$  by various researchers (Austroads, 2000; Broomfield, 2006; ACI 365.1R, 2010; Life-365, 2010).

In general, the value of  $C_0$  increases with the time of exposure of the structure. However, the data from the field indicate that  $C_0$  tends to reach a 'maximum' value, which generally depends on the concrete porosity and aggressiveness of chloride environment. In the present study, the surface chloride concentration is assumed to be 1% by weight of cement (similar to the typical values in the marine tidal zone, as in Life-365, 2010). Also, it should be noted that the threshold chloride concentration assumed is same for both OPC and PPC concretes, which may not be true (Richardson, 2002).

### Stable propagation period ( $t_{sp}$ )

Among the literature on service life prediction of concrete structures, the literature on corrosion propagation period is relatively less than that on the corrosion initiation period. In the present study,  $t_{sp}$  has been computed using the model proposed by Rodriguez et al. (1996). In this model, the loss in rebar diameter, at any time  $t$  (in years), can be estimated as:

$$\theta(t) = \theta(0) - p(t) \quad (3)$$

$$p(t) = 0.0116 \alpha i_{corr} (t_s - t_i) \quad (4)$$

where  $\theta(0)$  is the initial diameter of the rebar (in mm);  $p(t)$  is the loss of section (in mm) in time  $t$ ;  $t_{sp} = t_s - t_i$ ;  $i_{corr}$  is the average value of corrosion current density (in  $\mu\text{A}/\text{cm}^2$ ); 0.0116 is a factor which converts  $\mu\text{A}/\text{cm}^2$  to mm/year;  $t_s$  is the total time elapsed (in years),  $t_i$  is the time for initiation of corrosion (years), and  $\alpha$  is a pitting factor (for including the effect of highly localized pitting normally associated with chloride-induced corrosion; for uniform corrosion,  $\alpha = 2$  and for pitting corrosion,  $\alpha = 5$  to 10), as in González et al. (1995).

## ESTIMATION OF SERVICE LIFE

### Initiation period

For computing the corrosion initiation period,  $t_i$ , Eq. (2) can be written as:

$$t_i = \frac{x^2}{4D_c \left[ \text{erf}^{-1} \left( 1 - \frac{C_{tc}}{C_0} \right) \right]^2} \quad (5)$$

Using Eq. (5) and available values on diffusion coefficient,  $D_c$ , the corrosion initiation periods,  $t_i$ , have been evaluated for a cover thickness of 20 mm,  $C_{tc} = 0.4\%$  and  $C_0 = 1\%$ , by weight of cement, and are presented in Table 1. For OPC and PPC concretes, the  $D_c$  values

were assumed to be  $7.67 \times 10^{-13}$  and  $3.92 \times 10^{-13}$ , respectively. Based on these parametric assumptions, the corrosion initiation periods,  $t_i$ , for OPC and PPC concretes were estimated to be 2.1 and 4.2 years, respectively. It can be concluded that by using PPC instead of OPC, the corrosion initiation time can be approximately doubled.

### Stable propagation period

For the computation of  $t_{sp}$ , the corrosion rate or corrosion intensity level ( $i_{corr}$ ) and damage level are needed. As mentioned previously, the U-shaped specimens of OPC and PPC were subjected to accelerated chloride induced corrosion for the same duration (at 3.5% NaCl solution; 10 V anodic potential for 22 days). At the end of specified duration, specimens were autopsied and the rebars were removed and cleaned as per ASTM G1 to determine the extent of corrosion. The gravimetric weight loss (i.e., amount of corrosion) of the rebar was estimated as the loss in weight over the middle 300mm length of the rebar. Based on the gravimetric weight loss, the equivalent corrosion rate or current density,  $i_{corr}$ , for OPC and PPC concretes were estimated to be 768 and 289  $\mu\text{A}/\text{cm}^2$ , respectively (Bhaskar, 2012). It can be seen that the value for PPC concrete is much lower than that of OPC concrete due to the increased resistivity and lower porosity obtained by the incorporation of fly ash (Scott and Alexander, 2007; Bhaskar et al., 2011). Because these values have been obtained under accelerated conditions, they are likely to be much higher than what could occur in the field. As reported by Andrade et al. (1990) and Austin et al. (2004), the  $i_{corr}$  values obtained under field conditions for extreme chloride induced corrosion conditions are generally in the range of 100 to 200  $\mu\text{A}/\text{cm}^2$ , for normal grades of concrete (as in case of the OPC concrete of the present study). Andrade et al. (1990) and González et al. (1995) further reported that values of 1–3  $\mu\text{A}/\text{cm}^2$  are frequent in active corrosion and that values of the order of 10  $\mu\text{A}/\text{cm}^2$  or higher are seldom observed in the field.

For OPC concrete, the  $i_{corr}$  values obtained experimentally by Bhaskar (2013) and the typical range of values reported by Andrade et al. (1990) are significantly different. Therefore, it is necessary to map the laboratory  $i_{corr}$  value to the more realistic field value in order to estimate the service life using Eq. (5). Hence, in the present study, for OPC concrete, it is assumed that the  $i_{corr}$  value is between 1 and 10  $\mu\text{A}/\text{cm}^2$  (a more realistic field value) instead of  $i_{corr}$  values observed in laboratory (i.e., 768  $\mu\text{A}/\text{cm}^2$ ). The field value for PPC concrete is obtained from the ratio of laboratory  $i_{corr}$  values of PPC and OPC concrete as (Morinaga et al., 1994):

$$\left( \frac{i_{corr,OPC}}{i_{corr,PPC}} \right)_{Lab} = \left( \frac{i_{corr,OPC}}{i_{corr,PPC}} \right)_{Field} \quad (6a)$$

$$(i_{corr,PPC})_{Field} = \left( \frac{i_{corr,PPC}}{i_{corr,OPC}} \right)_{Lab} \times (i_{corr,OPC})_{Field} \quad (6b)$$

Now, another important parameter that is needed to estimate the stable propagation period ( $t_{sp}$ ) is the definition of the level or levels of deterioration that may affect the serviceability or the load-carrying capacity of the structure. The levels of deterioration classified in the Bulletin No. 162 (CEB, 1983) and Andrade et al. (1990) were considered in fixing the damage levels. The estimations in the present study are made for damage level C or for the upper limit of the loss in steel section of 10% (i.e., it is assumed that the end of stable propagation period ( $t_{sp}$ ) is reached when the steel section decreases by 10%).

Now, the last step in the estimation of  $t_{sp}$  consists of calculating the time taken to reach the assumed deterioration level, i.e., 10% loss in steel section from the initiation period. Therefore,

$$t_{sp} = \frac{\theta(0) - \theta(t)}{0.0116 \alpha i_{corr}} \quad (7)$$

where,

- $t_{sp}$  = stable propagation period (in years)
- $\theta(0)$  = initial diameter of the rebar (in mm)
- $\theta(t)$  = net diameter of the rebar (in mm), after an assumed 10% section loss
- $i_{corr}$  = average value of corrosion current density ( $\mu\text{A}/\text{cm}^2$ )
- $\alpha$  = pitting factor (for uniform corrosion,  $\alpha = 2$ ; pitting corrosion,  $\alpha = 5$  to 10)

Knowing the values of  $t_i$  and  $t_{sp}$ , the total service life ( $t_s$ ), can be estimated, using Eq. (1). Table 1 and Figure 3 present  $t_i$ ,  $t_{sp}$  and  $t_s$  for OPC and PPC concretes, when the field  $i_{corr}$ -value of OPC concrete is assumed as 1, 5 and  $10\mu\text{A}/\text{cm}^2$ . From these, it can be said that the service life of RC structures subjected to chloride induced corrosion depends largely on the chloride diffusion and aggressiveness of the exposure conditions. When the  $i_{corr}$  is  $1\mu\text{A}/\text{cm}^2$ , the service life ( $t_s$ ) of PPC concrete is approximately 2.5 times more than that of OPC concrete. When the  $i_{corr}$  is  $10\mu\text{A}/\text{cm}^2$ , the service life ( $t_s$ ) of PPC concrete is approximately 2 times more than that of OPC concrete. Life-365 (2010) takes  $t_{sp}$  as 6 years for most commonly used uncoated steels, which coincides with the value obtained here for OPC concrete with an  $i_{corr}$  of  $1\mu\text{A}/\text{cm}^2$ .

Table 1. Service lives of concretes for different field  $i_{corr}$ -values assumed for OPC concrete

Type of concrete	Corrosion initiation period ( $t_i$ ), years	Stable propagation period ( $t_{sp}$ ), years	Service life ( $t_s$ ), years
$i_{corr}$ -value of OPC concrete = $1\mu\text{A}/\text{cm}^2$			
OPC	2.1	6.0	8.1
PPC	4.2	16.3	20.5
$i_{corr}$ -value of OPC concrete = $5\mu\text{A}/\text{cm}^2$			
OPC	2.1	1.2	3.3
PPC	4.2	3.2	7.4
$i_{corr}$ -value of OPC concrete = $10\mu\text{A}/\text{cm}^2$			
OPC	2.1	0.6	2.7
PPC	4.2	1.6	5.8

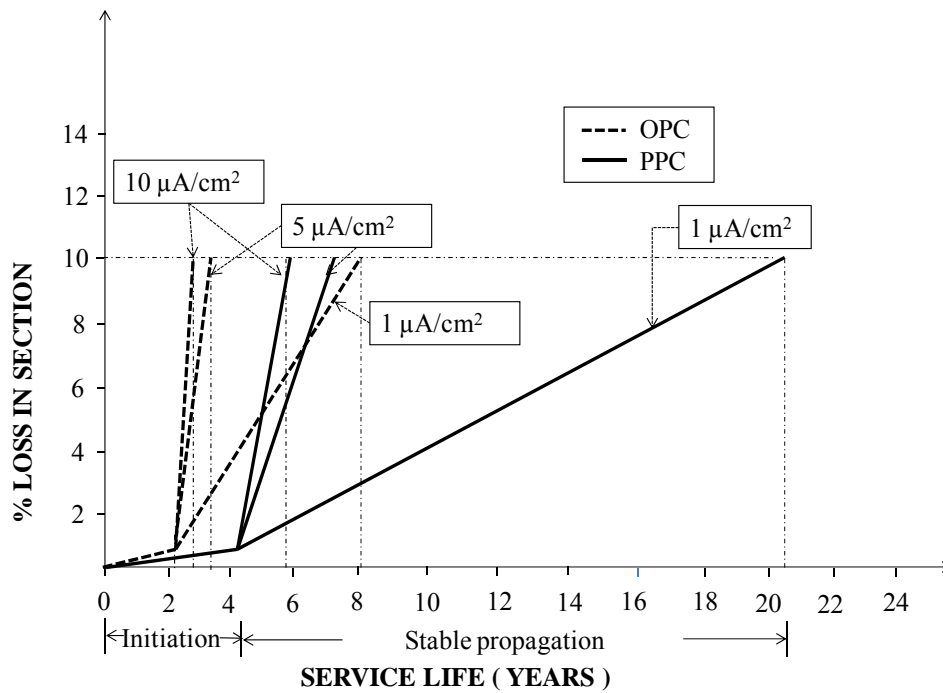


Figure 3. Service life of OPC and PPC concretes for different conditions

The service life estimations reveal that PPC concrete will have better corrosion resistance than OPC concrete due to the delayed corrosion initiation period and decreased corrosion rate during the propagation period. This can be attributed to the higher resistivity and pore refinement due to secondary hydration in the fly ash based PPC concrete. It can be observed that when corrosion rate is low, the durations of initiation and propagation periods are significant in estimating the overall service life; whereas when corrosion rate is high, the duration of initiation period is significant and the duration of propagation period is insignificant in estimating the overall service life.

As reported by Mehta (2002), the results obtained in this study also indicate that the use of PPC will prolong the useful life of a structure, and thereby contribute significantly to sustainable construction. In addition, the use of PPC will lead to the reduction of cement clinker usage, and consequently to minimising the consumption of limestone and other natural resources and to lowering the emission of CO<sub>2</sub>.

## CONCLUSIONS

The overall service life ( $t_s$ ) can be assumed as the sum of corrosion initiation period ( $t_i$ ) and stable propagation period ( $t_{sp}$ ). Based on this and under the conditions considered in this study, the following conclusions have been drawn:

- The  $t_i$  and  $t_{sp}$  values can be estimated based on the experimentally evaluated  $D_c$  and the computed  $i_{corr}$ -values, respectively. When the  $i_{corr}$ -values are low to medium, both  $t_i$  and  $t_{sp}$  contributes significantly to the overall service life. However, at higher  $i_{corr}$  (i.e., at severe chloride exposures), only  $t_i$  contributes significantly to the overall service life.
- The service life of PPC based concrete is found to be nearly twice that of corresponding OPC based concrete.



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## REFERENCES

- ACI 211.1 (2010) *Standard practice for selecting proportions for normal, heavyweight and mass concrete*, American Concrete Institute, USA.
- ACI 365.1R (2010) *Service life prediction – State of the art report*, American Concrete Institute, USA.
- Ahmad, S., Bhattacharjee, B., and Wason, R. (1997) Experimental service life prediction of rebar corroded reinforced concrete structure. *ACI Materials Journal*, 94 (4), 311-316.
- Alonso, C., Andrade, C., Castellote, M., and Castro, P. (2000) Chloride threshold values to depassivate reinforcing bars embedded in a standardized OPC mortar. *Cement and Concrete Research*, 30 (7), 1047-1055.
- Amey, S. L., Johnson, D. A., Miltenberger, M. A., and Farzam, H. (1998) Prediction of service life of concrete marine structures: An environmental methodology. *ACI Structural Journal*, 95 (2), 205-214.
- Andrade, C., Alonso, M. A., and Gonzalez, J. A. (1990) An initial effort to use the corrosion rate measurements for estimating rebar durability. pp. 29-37. In Berke, N.S., Chaker, V. and Whiting, D., (eds.) *Corrosion rates of steel in concrete*. ASTM STP 1065, Philadelphia, 1990.
- Angst, U., Elsener, B., Larsen, C. K., and Vennesland, Ø. (2009) Critical chloride content in reinforced concrete – A review. *Cement and Concrete Research*, 39, 1122-1138.
- Anoop, M. B., Balaji Rao, K. and Appa Rao, T. V. S. R. (2002) Application of fuzzy sets for estimating service life of reinforced concrete structural members in corrosive environments, *Engineering Structures*, 24 (9), 1229-1242.
- ASTM C 1202-97 (2009) Standard Test Method for electrical indication of concrete's ability to resist chloride penetration. *Annual book of ASTM standards*, American Society of Testing and Materials, USA.
- Austrroads (2000) *Service life prediction of reinforced concrete structures*, Austrroads Project No. N.T&E.9813, Austrroads Publication No. AP-T07/00, Sydney, Australia.
- Austin, S. A., Lyons, R., and Ing, M. J. (2004) Electrochemical behaviour of steel reinforced concrete during accelerated corrosion testing. *Corrosion*, 60 (2), 203-212.
- Balaji Rao, K., Anoop, M. B. Appa Rao, T. V. S. R. and Lakshmanan, N. (2008) Time variant reliability analysis of reinforced concrete members subjected to chloride induced corrosion of reinforcement, *Proceedings of Sixth Structural Engineering Convention, SEC-2008*, Structural Engineering Research Centre, Chennai, India, pp. 237-246.
- Bentur, A., Diamond, S., and Berke, S. (1998) *Steel corrosion in concrete: Fundamentals and civil engineering practice*, E & FN Spon.
- Bhaskar, S., (2013) Study of chloride induced corrosion of reinforcement steel in cracked concrete, *PhD Thesis, Indian Institute of Technology Madras*, Chennai, India.
- Bhaskar, S., Gettu, R., Bharatkumar, B. H., and Neelamegam, M. (2011) Chloride-induced corrosion of steel in cracked OPC and PPC concretes: Experimental study. *Journal of Materials in Civil Engineering, ASCE*, 23 (7), 1057-1066.

- Broomfield, J. P. (2006) *Corrosion of steel in concrete: understanding, investigation and repair*, 2<sup>nd</sup> Edition, Taylor & Francis Ltd, U.K.
- CEB Bulletin No. 162 (1983) Assessment of concrete structures and design procedures for upgrading, pp. 87-90.
- Gjørsv, O. E. (1994) Important test methods for evaluation of reinforced concrete durability. Pp. 545-574. In Mehta, P. K. (ed.) *Concrete in the 21<sup>st</sup> century: Past, present and future*. ACI SP 144, 1994.
- González, J. A., Andrade, C., Alonso, C., and Feliu, S. (1995) Comparison of rates of general corrosion and maximum pitting penetration on concrete embedded steel reinforcement. *Cement and Concrete Research*, 25 (2), 257-264.
- Life-365 *Service life prediction model and computer program for predicting the service life and life-cycle cost of reinforced concrete exposed to chlorides*, Version 2.0.1, Life-365 Consortium II, 2010.
- Mehta, P. K. (2002) Greening of the concrete industry for sustainable development, *Concrete Int.*, 24 (7), pp. 23-28.
- Morinaga, S., Irino, K., Ohta, T., and Arai, H. (1994) Life prediction of existing reinforced concrete structures determined by corrosion, pp.603-618. In Swamy R. N. (ed.), *Corrosion and Corrosion Protection of Steel in Concrete*, Sheffield, UK, Sheffield Academic Press, 1994.
- NT Build 355 (1997) Concrete, mortar and cement based repair materials: Chloride diffusion coefficient from migration cell experiments, Nordtest method.
- Pillai, R. G. (2003) Accelerated quantification of critical parameters for predicting the service life and life cycle costs of chloride-laden reinforced concrete structures, *MS Thesis*, Texas A&M University, College Station, Texas, USA.
- Pillai, R. G., and Trejo, D. (2005) Surface condition effects on critical chloride threshold of steel reinforcement. *ACI Materials Journal*, 102 (2), 103-109.
- Prabakar, J., Manoharan, P. D., and Chellappan, A. (2010) Diffusion characteristics of OPC concrete of various grades under accelerated test conditions. *Construction and Building Materials*, 24 (3), 346-352.
- Richardson, M. (2002) *Fundamentals of durable reinforced concrete*, Spon Press, London and New York.
- Rodriguez, J., Ortega, L. M., Casal, J., and Diez, J. M. (1996) Assessing structural conditions of concrete structures with corroded reinforcement. pp. 65-78. In Dhir R. K. and Jones M. R. (eds.), *Concrete Repair, Rehabilitation and Protection*, E & FN Spon, London, 1996.
- Saraswathy, V., and Song, H. W. (2007) Evaluation of corrosion resistance of Portland pozzolana cement and fly ash blended cements in pre-cracked reinforced concrete slabs under accelerated testing conditions. *Materials Chemistry and Physics*, 104, 356-361.
- Scott, A., and Alexander, M. G. (2007) The influence of binder type, cracking and cover on corrosion rates of steel in chloride contaminated concrete. *Magazine of Concrete Research*, 59 (7), 495-505.
- Tuutti, K. (1982) *Corrosion of Steel in Concrete*, CBI, Research Report 4, 1982, Stockholm.
- Zhang, T., and Gjørsv, O. E. (1994) An electrochemical method for accelerated testing of chloride diffusivity in concrete. *Cement and Concrete Research*, 24 (8), 1534-1548.